## Writing Chemical Formulas Level 2 WS

Review of All Ionic Compounds (including acids)

Write the correct chemical formula for each of the following compounds. For acids, include the teritary compound name. For ex. Chlorous acid – hydrogen chlorite –  $HclO_2$  You may not know all the metals listed, so look them up on the periodic table.

- 1. potassium sulfate
- 2. magnesium selenide
- 3. aluminum oxide
- 4. beryllium hydroxide
- 5. sodium nitrate
- 6. phosphoric acid
- 7. tin (IV) chloride
- 8. iron (III) sulfite
- 9. copper (I) sulfate
- 10. scandium (I) sulfide
- 11. ammonium fluoride
- 12. calcium sulfate
- 13. magnesium phosphate
- 14. acetic acid
- 15. sodium bisulfate
- 16. copper (II) nitrate
- 17. iron (III) sulfite
- 18. chromium (I) phosphide
- 19. strontium bicarbonate
- 20. ammonium phosphate